Total Synthesis and Stereostructure of Methyl (E)-O-Methylmulticolanate, a 4-Ylidenetetronic Acid from *Penicillium multicolor*

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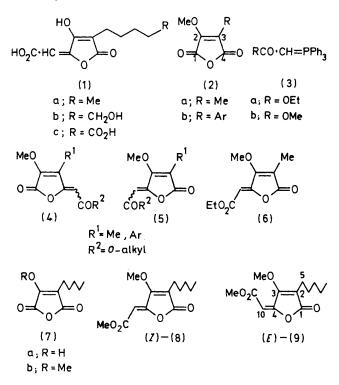
Summary A total synthesis of methyl (E)-O- methylmulticolanate (9), and its geometrical isomer (8), is described; the stereostructure of (9) was established by comparative spectral studies and X-ray analysis of the synthetic analogue (6).

MULTICOLANIC ACID (1a), and the related metabolites (1b) and (1c) comprise a novel group of 4-ylidenetetronic acids

isolated recently from cultures of *Penicillium multicolor*.¹† The detailed structure of multicolic acid (1b) was uniquely worked out largely from an examination of the ¹³C-couplings observed in the n.m.r. spectrum of the dimethyl derivative enriched with doubly labelled ¹³C acetate; the stereochemistry about the 4-ylidene bond was not determined. Here we report a total synthesis of the geometrical isomers of methyl *O*-methylmulticolanate (9), which confirms the

† Multicolanic acid (1a) has only recently been isolated from P. multicolor; personal communication from Dr. J. S. E. Holker.

overall structures assigned to the three natural metabolites (1a-c), and furthermore establishes their (E)-stereochemistry at the same time.



Earlier studies have shown that condensation reactions between 2-methoxy-substituted maleic anhydrides (2) and stabilised phosphoranes of type (3) provide flexible routes to (Z)-(E) mixtures of the corresponding ylidenebutenolide derivatives [(4) or (5)] resulting from regioselective addition, at either C-1 or C-4, in the anhydrides.² Although ¹H n.m.r. shift data and isomerisation studies permitted the assignment of (Z)- and (E)-geometry to the ylidene derivatives, they did not by themselves distinguish unambiguously the

- [‡] Details of the X-ray measurements will be published separately.
- ¹ J. A. Gudgeon, J. S. E. Holker, and T. J. Simpson, J.C.S. Chem. Comm., 1974, 636. ² M. J. Begley, D. W. Knight, and G. Pattenden, Tetrahedron Letters, 1975, 4279; D. W. Knight, Ph.D. Thesis, University of Nottingham, 1975. ³ F. Adickes and G. Andesen, Annalen, 1943, 555, 41.

 - ⁴ Cf. H. Schinz and M. Hinder, Helv. Chim. Acta., 1947, 30, 1349.

site of attack (C-1 vs. C-4) in the anhydrides. In our earlier work we assigned orientation (4) to these derivatives. Circumstances necessitated a re-examination of this proposal, and X-ray measurements on the (Z)-isomer, m.p. 66-67 °C, produced from the phosphorane (3a) and the anhydride (2a) now show that this molecule has the alternative orientation $(\mathbf{6})$; this observation suggested a simple route to the carbon skeleton in multicolanic acid (1a).

Condensation between ethyl heptanoate and diethyl oxalate, followed by treatment of the resulting oxalacetic ester³ with sulphuric acid led to the 2-hydroxymaleic anhydride (7a),4 m.p. 92-94 °C, vmax 1845, 1768, and 1698 cm⁻¹, which on methylation with diazomethane gave the 2-methoxy-derivative (7b). Reaction between the anhydride (7b) and the phosphorane (3b) then led to a 3:1 mixture of (Z)- and (E)-isomers (8) and (9), respectively, of methyl O-methylmulticolanate which was separated by chromatography. The structures and geometries of (8) and (9) followed from comparison between their spectral data and those of the 2-methyl analogue (6). The crystalline (Z)-isomer (8), m.p. 72-73 °C, is easily distinguished from the oily (E)-isomer by ¹H n.m.r. studies, since the olefinic hydrogen in (9) is considerably deshielded (τ 4.17 vs. 4.37) by virtue of its cis-relationship to the butenolide oxygen. The corresponding olefinic hydrogen absorptions in the dimethyl derivatives of the naturally derived 4-ylidenetetronic acids (1a—c) appear at τ ca. 4.17, which suggests that they all have the same (E)-stereochemistry. Confirmation of this assignment came from comparison of the ¹³C shift data of (9) [δ 161·1 (C-1), 110·6 (C-2), 168·7 (C-3), 150.9 (C-4), 22.3 (C-5), 297 (C-6), 23.5 (C-7), 31.6 (C-8), 13.9 (C-9), 101·1 (C-10), 164·4 (C-11), 59·7 (OMe), and 52·1 p.p.m. (OMe)] and of (8) [\$ 160.6, 109.7, 168.3, 150.2, 23.4, 29.7, 25.5, 32.2, 62.2, 100.9, 163.8, 59.5, and 52.0 p.p.m.], with those determined for naturally derived methyl O-methylmulticolanate.

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